

## 1,5-Asymmetric Induction in Addition Reaction of Aldehydes with Chiral Allyltitaniums Having an Amino Group at the Stereogenic Center. Synthesis of Optically Active 2,6-cis-Disubstituted Piperidines

Teng Xin, Sentaro Okamoto and Fumie Sato\*

Department of Biomolecular Engineering, Tokyo Institute of Technology, 4259 Nagatsuta-cho, Midori-ku, Yokohama, Kanagawa 226-8501, Japan

Received 11 June 1998; revised 9 July 1998; accepted 10 July 1998

## **Abstract**

Chiral allyltitaniums prepared from cyclic carbamate of optically active 4-aminoalk-1-en-3-ols and a Ti(O-i-Pr)4/2i-PrMgCl reagent react with aldehydes with good to excellent regio- and stereoselectivity to afford optically active 1,5-amino alcohols, from which optically active 2,6-cis-disubstituted piperidines are synthesized. © 1998 Elsevier Science Ltd. All rights reserved.

Keywords: titanium and compounds; asymmetric induction; amino alcohols; piperidines

Recently, we have developed a new efficient entry to allyltitanium compounds by the reaction of allylic alcohol derivatives such as halides, carbonates and acetates with a Ti(O-i-Pr)4/2i-PrMgCl reagent (1) which proceeds via an oxidative addition pathway [1]. The resulting allyltitaniums react selectively with aldehydes and imines, thus providing an efficient method for synthesizing homoallylic alcohols or amines [1,2]. In these previously reported examples, it was found that optically active 4-aminoalk-1-en-3-ols (2) reacted, after converting to the ethyl carbonate 3, with 1 to afford the corresponding chiral allyltitaniums and which in turn react with aldehydes regio- and stereoselectively to afford  $\beta$ -vinyl- $\gamma$ -aminoalkanols 4 (R = Bn or Boc) having the structure depicted in Scheme 1 [2d]. We have

	Substrate	es	Product(s) <sup>b</sup> OH  R <sup>1</sup> $R^2$ Bn-NH $6^d$		
Entry	R <sup>1</sup> O Bri O 5°	R <sup>2</sup> CHO			
			<del>*************************************</del>	Diastereoselectivity <sup>e</sup>	Yield, % <sup>f</sup>
1	$R^1 = CH_3$	n-PrCHO	6a	5 : 1 <sup>g</sup>	81
2	$R^1 = n$ -Pr	EtCHO	6b	14:1	79
3	$R^1 = CH_2OTBS$	MeCHO	6c	14:1	83
4		EtCHO	6d	18 : 1 <sup>9</sup>	82
5		PhCHO	6e	20 : 1	79
6		CHO	6f	11:1	81

Table 1. The Addition Reaction of Allyltitaniums Derived from 5 with Aldehydes<sup>a</sup>

<sup>a</sup>A mixture of **5** (1.0 equiv), Ti(O-i-Pr)<sub>4</sub> (1.5 equiv), and i-PrMgCl (3.0 equiv) in ether was stirred for 1.5 h at -50~-40 °C, and then the mixture was allowed to warm to 0 °C and stirred for 2 h. After cooling to -78 °C, to this was added an aldehyde (1.5 equiv). <sup>b</sup>The reaction afforded a mixture of **6** and its regioisomer **4** (R = H) in a ratio of more than 12:1 in all cases. The ratio was determined by <sup>1</sup>H and/or <sup>13</sup>C NMR. <sup>c</sup>5 was prepared from **2** (R=Boc) by treatment with NaH in THF (88~92% yield). A mixture of two diastereomers in a ratio of 4:1 for entry 1 and entry 2, 7:3 for entries 3-6 was used. <sup>d</sup>>98% (Z)-olefin geometry in all cases. <sup>e</sup>Determined by <sup>1</sup>H and/or <sup>13</sup>C NMR analysis after separation of **4**. <sup>f</sup>Combined yield of two diastereomers of **6**. <sup>g</sup>Stereochemistry was confirmed by converting to the corresponding piperidine, see Scheme 2.

now found that allyltitaniums derived from cyclic carbamates 5 (prepared easily from 2<sup>1</sup> (R=Boc) by treatment with NaH) react with aldehydes to afford another regioisomer, 1,5-amino alcohols 6, highly selectively (Scheme 1, Table 1).

As shown in Table 1, the reaction of the allyltitaniums generated in situ from 5 and 1 with aldehydes afforded 1,5-amino alcohols 6 and their regioisomers 4 (R = H) in a ratio of more than 12:1. The 1,5-amino alcohols 6 thus produced consisted of two diastereomers having Z-geometry with respect to the olefin moiety; the isomer having E-olefin geometry was not detected. The diastereoselectivity associated with the 1,5-asymmetric induction was somewhat dependent on the steric bulkiness of  $R^1$  in 5 and, when  $R^1$  is a methyl group, it was 5:1 (entry 1) while it was better than 11:1 in the case where the bulkiness of  $R^1$  is larger than that of the methyl group. The stereochemistry of the main diastereoisomer was verified as shown in Table 1 by  $^1H$  NMR analysis after derivatization to the corresponding piperidine derivative in several representative cases (vide infra). It should be noted that in these reactions, although the regioisomers 4 and 6 could be readily separated from each

$$R = CH_3 \text{ or } CH_2OH$$

$$R = CH_3 \text{ or } CH_2OH$$

$$R = CH_3 \text{ or } CH_2OH$$

$$R = CH_3 \text{ or } CH_2OHS$$

$$R = CH_3 \text{ or } CH_2OTBS$$

$$R = CH_3 \text{ or } CH_2OTBS$$

<sup>&</sup>lt;sup>1</sup>The compound 2 (>94% e.e.) where  $R^1$  is CH<sub>3</sub> or CH<sub>2</sub>OTBS was synthesized from the corresponding L-amino acid by the conventional reaction sequences shown below. Meanwhile, the compound 2 where  $R^1$  is n-Pr (95% e.e.) was prepared from readily available optically active 2,3-epoxy-1-hexanol: see ref. 2d.

other by column chromatography, two stereoisomers of 6 were inseparable. However, since the diastereoselectivity is very high except for the case where  $R^1$  is a methyl group, the value of the reaction as synthetic methodology of 6 would not be detracted<sup>2</sup>.

With a convenient method for synthesizing optically active 6 in hand, we carried out their conversion into piperidine derivatives in several representative cases. Thus, as illustrated in Scheme 2, 2,6-disubstituted 3,4-didehydropiperidines (1,2,5,6-tetrahydropyridines) 7a and 7d³ were synthesized in good overall yields from the corresponding 6 by the following reaction sequence: (1) protection of the amino group with Boc using (Boc)<sub>2</sub>O and Et<sub>3</sub>N in THF at room temperature, (2) removal of the benzyl group using Na / liq. NH3 at -70 ~ -40 °C, (3) mesylation of the alcohol group using MsCl and Et<sub>3</sub>N in CH<sub>2</sub>Cl<sub>2</sub> at ambient temperature, and (4) cyclization reaction using *t*-BuOK in THF at 0 °C. The unsaturated piperidines 7 thus obtained were hydrogenated (H<sub>2</sub>, Pd/C, EtOH, room temp.) quantitatively to saturated 2,6-disubstituted piperidines 8, the *cis*-stereochemistry of which was confirmed by  $^{1}$ H NMR analysis<sup>4</sup>. It should be noted that a new asymmetric entry has now been opened up to saturated and unsaturated *cis*-2,6-disubstituted piperidines, the structure of which frequently occurs as a subunit of alkaloids and biologically active compounds [3,4].

The regio- and stereochemistry of the aldehyde addition reaction of the allyltitanium derived from 5 and 1 can be explained by the preference for the secondary allyltitanium 10

<sup>&</sup>lt;sup>2</sup>The characteristic <sup>1</sup>H NMR data of 6 (300MHz, CDCl<sub>3</sub>): for 6a: δ 1.17 (d, J = 6.0 Hz, 0.5H, CH<sub>3</sub>CHN, minor), 1.18 (d, J = 6.3 Hz, 2.5H, CH<sub>3</sub>CHN, major), 3.51-3.68 (m, 2H, CHN, CHOH), 5.41-5.63 (m, 2H, CH=CHCH<sub>2</sub>). for 6b: δ 3.36-3.46 (m, 1H, CHN), 3.51 (quintet, J = 6.0 Hz, 1H, CHOH), 5.41 (t, J = 10.2 Hz, 1H, CH=CHCH<sub>2</sub>), 5.63 (dt, J = 10.8, 7.8 Hz, 1H, CH=CHCH<sub>2</sub>). for 6c: δ 3.50-3.85 (m, 4H, TBSOCH<sub>2</sub>CHN, CHOH), 5.43-5.52 (m, 1H, CH=CHCH<sub>2</sub>), 5.68 (dt, J = 7.8, 11.1 Hz, 1H, CH=CHCH<sub>2</sub>). for 6d: δ 3.45-3.65 (m, 4H, TBSOCH<sub>2</sub>CHN, CHOH), 5.42-5.51 (m, 1H, CH=CHCH<sub>2</sub>), 5.68 (dt, J = 7.8, 11.1 Hz, 1H, CH=CHCH<sub>2</sub>). for 6e: δ 3.43-3.61 (m, 3H, TBSOCH<sub>2</sub>CHN), 4.68 (dd, J = 4.5, 8.1 Hz, 1H, CHOH), 5.52 (dd, J = 7.8, 10.8 Hz, 1H, CH=CHCH<sub>2</sub>), 5.70 (ddd, J = 7.8, 8.4, 10.8 Hz, 1H, CH=CHCH<sub>2</sub>). for 6f: δ 3.48-3.65 (m, 3H, TBSOCH<sub>2</sub>, CHN), 4.04 (q, J = 6.3 Hz, 1H, CHOH), 5.40-5.55 (m, 2H, CH=CHCH<sub>2</sub>, CH=CHCH<sub>3</sub>), 5.60-5.73 (m, 2H, CH=CHCH<sub>2</sub>, CH=CHCH<sub>3</sub>). The characteristic <sup>1</sup>H NMR data of cis-7 (300 MHz, CDCl<sub>3</sub>, 65 °C): For 7a δ 1.93 (dd, J = 6.3, 17.1 Hz, 1H, CH=CHCH<sub>2</sub>), 2.23-2.37 (m, 1H, CH=CHCH<sub>2</sub>), 4.28-4.43 (m, 2H, 2CHN), 5.54 (dt, J = 10.2, 3.3 Hz, 1H, olefinic proton), 5.63-5.74 (m, 1H, OCH<sub>2</sub>), 3.78 (dd, J = 4.5, 9.0 Hz, 1H, OCH<sub>2</sub>), 4.18-4.38 (m, 2H, 2CHN), 5.70-5.79 (m, 1H, olefinic proton), 5.88 (dt, J = 10.2, 3.0 Hz, 1H, olefinic proton).

<sup>&</sup>lt;sup>4</sup>The *cis*- and *trans*-7 (or 8) were inseparable by column chromatography. The ratios determined by  $^{1}$ H NMR analysis were as follows: for 7a or 8a: *cis*: *trans* = 4.8: 1; for 7d or 8d: *cis*: *trans* = 11:1. The *cis*-stereochemistry of the major isomer of 8a was confirmed by comparing the NMR data with those reported in the literature (Beak P, Lee WK. J. Org. Chem. 1993;58:1109-1117). The *cis*-stereochemistry of 8d was confirmed by  $^{1}$ H NMR NOE-difference experiments between protons at the C-2 and C-6 position of the piperidine ring. The characteristic  $^{1}$ H NMR data of 8d (300 MHz, CDCl<sub>3</sub>, 65  $^{\circ}$ C): δ 3.50 (dd, J = 5.1, 9.6 Hz, 1H, OCH<sub>2</sub>), 3.57 (t, J = 9.6 Hz, 1H, OCH<sub>2</sub>), 3.92-4.04 (m, 1H, CH<sub>3</sub>CH<sub>2</sub>CH<sub>N</sub>), 4.09-4.21 (m, 1H, OCH<sub>2</sub>CH<sub>N</sub>).

over the primary ones 9 and its reaction with aldehydes through a six-membered chair-like transition state as shown in Scheme 3. Usually, substituted allyltitanium exists as the primary titanium derivative over the more congested secondary ones which provided a basis for the production of 4 from the reaction of the allyltitaniums derived from 3 and 1 with aldehydes (see Scheme 1) [2d]; however, in the present case the equilibrium might be shifted in favor of 10 because it has a stable six-membered cyclic structure [5].

It should be noted that there is a precedent for highly selective 1,5-asymmetric induction in the addition reaction of aldehydes with allylmetal complexes having an amino group at the stereogenic center; Thomas *et al.* have reported that the reaction of tributyl(4-dibenzylaminopent-2-enyl)stannane with aldehydes proceeds with excellent selectivity to afford 1,5-amino alcohols having a similar structure to that of 6 [6]. They explained the stereoselectivity of the reaction by a similar transition state model to that shown in Scheme 3.

## References

- [1] Kasatkin A, Nakagawa T, Okamoto S, Sato F. J. Am. Chem. Soc. 1995;117:3881-3882.
- [2] (a) Gao Y, Sato F. J. Org. Chem. 1995;60:8136-8137. (b) Kasatkin A, Sato F. Angew. Chem., Int. Ed. Engl. 1996;35:2848-2849. (c) Hikichi S, Gao Y, Sato F. Tetrahedron Lett. 1997;38:2867-2870. (d) Teng X, Kasatkin A, Kawanaka Y, Okamoto S, Sato F. Tetrahedron Lett. 1997;38:8977-8980.
- [3] For recent reviews on natural products having a cis-2,6-disubstituted piperidine moiety, see: Numata A, Ibuka I. The Alkaloids, Brossi A. editor, New York, Academic Press: 1987;Vol. 31:193-315. Takahata H, Momose T. The Alkaloids, Cordell GA. editor, San Diego, Academic Press, 1993;Vol. 44:189-256. For recent syntheses of optically active cis-2,6-disubstituted piperidines, see: Munchhof MJ, Meyers AI. J. Am. Chem. Soc. 1995;117:5399-5400. Hirai Y, Watanabe J, Nozaki T, Yokoyama H, Yamaguchi S. J. Org. Chem. 1997;62:776-777. Momose T, Toyooka N, Jin M. J. Chem. Soc., Perkin Trans. 1, 1997; 2005-2013 and references cited therein.
- [4] For representative examples of natural products having a 2,6-disubstituted tetrahydropyridine moiety, see: Lotter HL, Abraham DJ, Turner CE, Knapp JE, Schiff PL, Slatkin DJ. Tetrahedron Lett. 1975; 2815-2818. Nader B, Bailey TR, Franck RW, Weinreb SM. J. Am. Chem. Soc. 1981;103:7573-7580. Colau B, Hootelé C. Can. J. Chem. 1983;61:470-472. Wasserman HH, Leadbetter MR, Kopka IE. Tetrahedron Lett. 1984;25:2391-2394. Natsume M, Ogawa M. Chem. Pharm. Bull. 1984;32:3789-3791. Bailey TR, Garigipati RS, Morton JA, Weinreb SM. J. Am. Chem. Soc. 1984;106:3240-3245. For syntheses of 2,6-disubstituted tetrahydropyridines, see: Bonin M, Romero JR, Grierson DS, Husson HP. J. Org. Chem. 1984;49:2392-2400. Wuts PGM, Jung YW. J. Org. Chem. 1988;53:1957-1965. Comins DL, Weglarz MA. J. Org. Chem. 1991;56:2506-2512. Ahman J, Somfai P. J. Am. Chem. Soc. 1994;116:9781-9782. Huwe CM, Velder J, Blechert S. Angew. Chem., Int. Ed. Engl. 1996;35:2376-2378. Bubnov YN, Klimkina EV, Ignatenko AV, Gridnev ID. Tetrahedron Lett. 1997;38:4631-4634. Craig D, McCague R, Potter GA, Williams MRV. Synlett, 1998;55-57 and references cited therein.
- [5] Hanko R, Hoppe D. Angew. Chem. 1982;94:378-379. Weidmann B, Seebach D. Angew. Chem. 1983;95:12-26. Zubaidha PK, Kasatkin A, Sato F. Chem. Commun. 1996;197-198.
- [6] Stanway SJ, Thomas EJ. J. Chem. Soc., Chem. Commun. 1994;285-286.